# A nonlinear coupling of chemical kinetics with mechanics

#### V. Klika<sup>1,2</sup>, M. Grmela<sup>3</sup>

<sup>1</sup>Institute of Thermomechanics of the AS CR, CR klika@it.cas.cz
<sup>2</sup>Dept. of Mathematics, FNSPE, Czech Technical University in Prague, CR <sup>3</sup>Chemical Engineering, Ecole Polytechnique, Montreal, Canada

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#### Outline



- 2 A first approximation of coupling CIT
- 3 A non-linear mechano-chemical coupling GENERIC
- 4 A feedback to motivation?



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#### Motivation

#### Bone and BR



# Why are we interested?

- the importance of mechanics (coupling)
- two groups of models (lack of communication)
- bridging them with NET

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Functions of BR in bone:to keep bone alive, to alter the shape of bone, repair damages in bone tissue, part of metabolism

IWNET 12 3 / 25

### Finding model formulation I

For interaction schemes

$$\sum_{j} \nu_{\alpha j} N_{j} \stackrel{k_{\pm \alpha}}{\rightleftharpoons} \sum_{j} \nu_{\alpha j}^{'} N_{j}, \quad \alpha = 1, .., s$$

the Law of Mass Action is used (with some limitations)

$$r_{\alpha} = k_{+\alpha} \prod_{j=1}^{n} [N_j]^{\nu_{\alpha j}} - k_{-\alpha} \prod_{j=1}^{n} [N_j]^{\nu'_{\alpha j}}$$

• Consequently, the change in concetration of subst. in time

$$[\dot{\mathrm{N}}_{j}] = \sum_{lpha} (
u_{lpha j} - 
u_{lpha j}^{'}) r_{lpha}$$

where *j* refers to a substance  $N_j$ ,  $r_\alpha$  is the rate of  $\alpha - th$  interaction,  $\nu_{\alpha j}$  is stoichiometric coefficient for entering, outcomming substances  $N_i$  of interaction  $\alpha$ , respectively

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IWNET 12 4 / 25

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# Finding model formulation of II

Phenomenological relations

#### Entropy production

$$\sigma(S) = \mathbf{j}_q \nabla \frac{1}{T} - \sum_{i=1}^n \mathbf{j}_{\mathrm{D}i} \left( \nabla \frac{\mu_i}{T} - \frac{\mathbf{F}_i}{T} \right) + \frac{1}{T} \mathbf{t}_{\mathrm{dis}} : \nabla \mathbf{v} + \frac{1}{T} \sum_{\alpha=1}^s r_\alpha \mathcal{A}_\alpha \ge 0.$$

and its general form:

$$\sigma(S) = J_s X_s + \mathbf{J}_v \cdot \mathbf{X}_v + \mathbf{J}_a^a \cdot \mathbf{X}_a^a + \mathbf{J}_t : \mathbf{X}_t.$$

CIT for a scalar (rate of chemical reaction):

$$J_{s} = L_{ss}X_{s} + \mathbf{L}_{sv} \cdot \mathbf{X}_{v} + \mathbf{L}_{sa}^{a} \cdot \mathbf{X}_{a}^{a} + \mathbf{L}_{st(s)}^{\circ} : \overset{\circ}{\mathbf{X}}_{t(s)}^{\circ}.$$

WLOG  $L_{st}$  is of the same kind as the thermodynamic force  $X_t$  (thus,  $L_{st} = L_{st(s)}^{\circ}$  is a symmetric tensor with zero trace) Rate of deformation tensor and entropy production

$$\mathbf{D}^{ij} = \dot{\mathbf{E}}^{ij} = \frac{1}{2} \left( \frac{\partial \dot{\mathbf{u}}^i}{\partial x^j} + \frac{\partial \dot{\mathbf{u}}^j}{\partial x^i} \right) = \frac{1}{2} \left( \frac{\partial \mathbf{v}^i}{\partial x^j} + \frac{\partial \mathbf{v}^j}{\partial x^i} \right) = \frac{1}{2} \left( (\nabla \mathbf{v})^{ij} + (\nabla \mathbf{v})^{ji} \right),$$

and for the rate of volume variation  $D^{(1)}$  it holds:

$$\mathbf{D}^{(1)} = \dot{\mathbf{E}}^{(1)} = \operatorname{div} \mathbf{v} = -\frac{1}{\rho} \frac{\mathrm{d}\rho}{\mathrm{d}t}$$

note: tensor  $\boldsymbol{\mathsf{D}}$  is the symmetric part of tensor  $\boldsymbol{\nabla}\boldsymbol{\mathsf{v}}$  and thus

$$T\sigma(S) = \mathbf{t}_{\mathrm{dis}} : \nabla \mathbf{v} + \sum_{\alpha=1}^{s} r_{\alpha} \mathcal{A}_{\alpha} =$$
$$= \left(\sum_{\alpha=1}^{s} r_{\alpha} \mathcal{A}_{\alpha} + \frac{1}{3} \mathrm{tr} \left(\nabla \mathbf{v}\right) \mathrm{tr} \left(\mathbf{t}_{\mathrm{dis}}\right)\right) + \left(\nabla \mathbf{v}\right)_{(s)} \cdot \left(\mathbf{t}_{\mathrm{dis}}\right)_{(s)} + \overset{\circ}{\mathbf{D}} : \left(\overset{\circ}{\mathbf{t}}_{\mathrm{dis}}\right)_{(s)} \ge 0,$$

where  $\nabla \mathbf{v}$  was decomposed into scaled unit tensor and a symmetric and an antisymmetric parts with zero traces.

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Chemo-Mechanical Coupling

IWNET 12 6 / 25

#### The isotropic system

To recall:

$$J_{s} = L_{ss}X_{s} + \mathbf{L}_{sv} \cdot \mathbf{X}_{v} + \mathbf{L}_{sa}^{a} \cdot \mathbf{X}_{a}^{a} + \mathbf{L}_{st(s)}^{\circ} : \overset{\circ}{\mathbf{X}}_{t(s)}^{\circ}.$$

Constraints on phenomenological coefficients follow (from invariance of isotropic system under orthogonal transformations - does not modify the phenomenological tensors): inversion ( $\mathbf{L}_{sv} = 0$ ), arbitrary rotation ( $\mathbf{L}_{sa}^a = 0$ )

Finally, from the fact that scalar quantity is not affected by orthogonal transf, namely  $\mathbf{a}^T \mathbf{b}$  it follows

$$L_{st} = L_{st}U.$$

 $\begin{array}{ll} \mbox{Moreover, } {\rm tr} \mbox{ } L_{st} = 0 & \Rightarrow & L_{st} = 0. \\ \mbox{In total (Curie principle):} \end{array}$ 

$$J_s = L_{ss}X_s$$

or in particular

$$r_{\alpha} = L_{s_1} \mathcal{A}_{\alpha} + L_{s_2} \mathbf{D}^{(1)}.$$

### Other choices of decomposition - scalar quantity $\mathbf{D}^{(2)}$

Analogous procedure; difference - tensors are not traceless Still constraints on phenomenological coefficients follow (from invariance of isotropic system under orthogonal transformations - does not modify the phenomenological tensors): inversion ( $L_{sv} = 0$ ), arbitrary rotation ( $L_{sa}^a = 0$ ) but  $L_{st} \neq 0$ , only

$$\mathbf{L_{st}} = L_{st} \mathbf{U},$$

In total, CIT and Curie-Prigogine principle leads to (notice the dynamic origin; static I. when viscous effects are significant but still through strain rate)

$$r_{\alpha} = L_{s_1} \mathcal{A}_{\alpha} + \left\lfloor L_{s_2} \mathbf{D}^{(1)} \right\rfloor$$
$$= L_{s_1} \mathcal{A}_{\alpha} + L_{s_2} \mathbf{D}^{(2)} + \left\lfloor L_{s_3} \mathbf{D}^{(1)} \right\rfloor$$

Klika, J Phys Chem B. 2010

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# Finding model formulation III

Modified Law of Mass Action

The most important (least worst option) invariant is  $\mathbf{D}^{(1)}$  and thus

$$\implies$$
  $r_{\alpha} = I_{\alpha\alpha} \mathcal{A}_{\alpha} + I_{\alpha\nu} \mathbf{D}^{(1)}$ 

Expressions of affinity and chemical potential in CIT lead to The Law of Mass Action (when coupling neglected)

$$r_{\alpha} = l_{\alpha\alpha} \mathcal{A}_{\alpha} = k_{+\alpha} \prod_{j=1}^{n} [N_{j}]^{\nu_{\alpha i}} - k_{-\alpha} \prod_{j=1}^{n} [N_{j}]^{\nu'_{\alpha i}}$$

The modified Law of Mass Action (including mechano-chemical coupling)

$$r_{\alpha} = l_{\alpha\alpha}\mathcal{A}_{\alpha} + l_{\alpha\nu}\mathbf{D}^{(1)} = k_{+\alpha}\prod_{j=1}^{n}[N_{j}]^{\nu_{\alpha i}} - k_{-\alpha}\prod_{i=1}^{n}[N_{j}]^{\nu_{\alpha i}'} + \boxed{l_{\alpha\nu}\mathbf{D}^{(1)}}$$

$$k_{\alpha}^{i} = C^{ijkl}(N_{j}) = C^{ijkl}(N_{j}(\mathbf{D}^{(1)}))$$

$$k_{\alpha}^{i} = L^{ijkl}(N_{j}) = C^{ijkl}(N_{j}(\mathbf{D}^{(1)}))$$

Klika, Maršík, *J Phys Chem B*. 2009 Klika,Grmela ()

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IWNET 12 9 / 25

#### Prerequisities

- Isothermal  $\phi$  instead of S
- Mechanical *scalar* variables (coupling depends on parity): even-parity **a**, and odd-parity **b**
- $\bullet$  Chemical variables: molar concentrations  ${\bf n}$  and their fluxes  ${\bf z}$
- Conjugates ( $\mathbf{a}^*, \mathbf{b}^*$ ), ( $\mathbf{n}^*, \mathbf{z}^*$ ),  $\mathbf{a}^* = \partial_{\mathbf{a}} \phi$

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#### Uncoupled mechanics

The time evolution of simplified but complete mechanics relevant for coupling (GENERIC):

$$\begin{pmatrix} \dot{\mathbf{a}} \\ \dot{\mathbf{b}} \end{pmatrix} = \begin{pmatrix} 0 & \kappa \\ -\kappa^{\mathsf{T}} & 0 \end{pmatrix} \begin{pmatrix} \mathbf{a}^{*} \\ \mathbf{b}^{*} \end{pmatrix} - \begin{pmatrix} 0 \\ \Theta_{\mathbf{b}^{*}}^{mech} \end{pmatrix}$$
$$= \begin{pmatrix} \kappa \mathbf{b}^{*} \\ -\kappa^{\mathsf{T}} \mathbf{a}^{*} \end{pmatrix} - \begin{pmatrix} 0 \\ \Theta_{\mathbf{b}^{*}}^{mech} \end{pmatrix}.$$

The dissipation potential  $\Theta^{mech}(\mathbf{a}, \mathbf{b}^*)$  satisfies:

$$\begin{split} &\Theta^{mech} \text{ is a real valued and sufficiently regular function of } (\mathbf{a}, \mathbf{b}, \mathbf{b}^*) \\ &\Theta^{mech}(\mathbf{a}, \mathbf{b}, 0) = 0 \\ &\Theta^{mech}(\mathbf{a}, \mathbf{b}, \mathbf{b}^*) \text{ reaches its minimum at } \mathbf{b}^* = 0 \\ &\Theta^{mech}(\mathbf{a}, \mathbf{b}, \mathbf{b}^*) \text{ is a convex function of } \mathbf{b}^* \text{ in a neighbourhood of } \mathbf{b}^* = 0. \end{split}$$

 $\Theta^{mech}$  is usually considered quadratic.

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IWNET 12 11 / 25

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Notes on structure of the evolution eq

$$\begin{pmatrix} \dot{\mathbf{a}} \\ \dot{\mathbf{b}} \end{pmatrix} = \begin{pmatrix} 0 & \boldsymbol{\kappa} \\ -\boldsymbol{\kappa}^{T} & 0 \end{pmatrix} \begin{pmatrix} \mathbf{a}^{*} \\ \mathbf{b}^{*} \end{pmatrix} - \begin{pmatrix} 0 \\ \Theta_{\mathbf{b}^{*}}^{mech} \end{pmatrix}$$
$$= \begin{pmatrix} \boldsymbol{\kappa}\mathbf{b}^{*} \\ -\boldsymbol{\kappa}^{T}\mathbf{a}^{*} \end{pmatrix} - \begin{pmatrix} 0 \\ \Theta_{\mathbf{b}^{*}}^{mech} \end{pmatrix},$$

**The first term**: time reversibility (invariance under  $t \rightarrow -t$  and change of parity).

**The second term**: in mechanics, dissipation is typically a friction  $\propto$  lin momenta  $\mathbf{b} \Rightarrow$  dissipation term has odd-parity;  $\Rightarrow$  dissipation term is expected to be only in the evolution of linear momentum  $\mathbf{b}$ ; We let only the odd-parity variables dissipate

Note that  $\frac{d\Phi}{dt} \leq 0$  (consistency with thermodynamics).

#### Uncoupled chemisty, LoMA

LoMA (Guldberg-Waage) in GENERIC form (yesterday):

$$\dot{\mathbf{n}} = -\Xi_{\mathbf{n}^*}^{chem}$$

with dissipation potential

$$arepsilon^{chem}(\mathbf{n},\mathbf{X}) = \sum_{
ho=1}^{s} W_{
ho}(\mathbf{n}) \left(e^{-rac{1}{2}X_{
ho}} + e^{rac{1}{2}X_{
ho}} - 2
ight)$$

where thermodynamic forces  $\mathbf{X} = (X_1, ..., X_s)^T$  are

$$X_{\rho} = \sum_{k=1}^{K} \gamma_{\rho}^{k} \Phi_{n_{k}}^{chem}$$

together with the natural choice of entropy form  $S(\mathbf{n}) = -\sum_{j=1}^{K} (n_j \ln n_j + Q_j n_j)$ where  $Q_1, ..., Q_K$  can be calculated. Note that  $\mathbf{n}^* = \mu$ .

IWNET 12 13 / 25

### Uncoupled chemisty, Extended

Fluxes (odd-parity momentum-like quantity) considered as independent variables (motivated by EIT; yesterday) From the analogy with mechanics

$$\begin{pmatrix} \dot{\mathbf{n}} \\ \dot{\mathbf{z}} \end{pmatrix} = \begin{pmatrix} 0 & \gamma \\ -\gamma^{T} & 0 \end{pmatrix} \begin{pmatrix} \mathbf{n}^{*} \\ \mathbf{z}^{*} \end{pmatrix} - \begin{pmatrix} 0 \\ \Theta_{\mathbf{z}^{*}}^{chem} \end{pmatrix}$$
$$= \begin{pmatrix} \gamma \mathbf{z}^{*} \\ -\gamma^{T} \mathbf{n}^{*} \end{pmatrix} - \begin{pmatrix} 0 \\ \Theta_{\mathbf{z}^{*}}^{chem}, \end{pmatrix}$$

Note that the reversible part of the right-hand-side of  $\dot{z}$  is equal to affinity. The standard LoMA formulation in GENERIC is required when fluxes equilibrate (separation of timescales)  $\rightarrow$  disip pot  $\Theta^{chem}$ . Now **n** acquire dissipation.

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#### Coupled chemical kinetics and mechanics

$$\begin{pmatrix} \dot{\mathbf{n}} \\ \dot{\mathbf{a}} \\ \dot{\mathbf{z}} \\ \dot{\mathbf{b}} \end{pmatrix} = \begin{pmatrix} 0 & 0 & \gamma & \mu \\ 0 & 0 & \nu & \kappa \\ -\gamma^{T} & -\nu^{T} & 0 & 0 \\ -\mu^{T} & -\kappa^{T} & 0 & 0 \end{pmatrix} \begin{pmatrix} \mathbf{n}^{*} \\ \mathbf{a}^{*} \\ \mathbf{z}^{*} \\ \mathbf{b}^{*} \end{pmatrix} - \begin{pmatrix} 0 \\ 0 \\ \Theta_{\mathbf{z}^{*},n,a,z,b)} \\ \Theta_{\mathbf{b}^{*}}^{(1,n,a,z,b)} \\ \Theta_{\mathbf{b}^{*}}^{(2,n,a,z,b)} \end{pmatrix}$$
$$= \begin{pmatrix} \gamma \mathbf{z}^{*} + \mu \mathbf{b}^{*} \\ \kappa \mathbf{b}^{*} + \nu \mathbf{z}^{*} \\ -\gamma^{T} \mathbf{n}^{*} - \nu^{T} \mathbf{a}^{*} - \Theta_{\mathbf{z}^{*}}^{(1,n,a,z,b)} \\ -\kappa^{T} \mathbf{a}^{*} - \mu^{T} \mathbf{n}^{*} - \Theta_{\mathbf{b}^{*}}^{(2,n,a,z,b)} \end{pmatrix}$$

Again dissipation is assumed only in evolution of odd-parity variables and reversible evolution of a state variable is caused by conjugate state variables with different parity

### Coupling, separation of scales

Relaxation times of reaction fluxes  $\ll$  relaxation times of concentrations, i.e.  $\dot{z}=0$  with solution  $z^*=Z^*(n,a,b)$  (we aim for modified LoMA). The time evolution equations become now

$$\begin{pmatrix} \dot{\mathbf{n}} \\ \dot{\mathbf{a}} \\ \dot{\mathbf{b}} \end{pmatrix} = \begin{pmatrix} \gamma \mathbf{Z}^* + \mu \mathbf{b}^* \\ \kappa \mathbf{b}^* + \nu \mathbf{Z}^* \\ -\kappa^T \mathbf{a}^* - \mu^T \mathbf{n}^* - \Theta_{\mathbf{b}^*}^{(2,n,a,Z,b)} \end{pmatrix}$$

Note the linear dependence of  $\dot{\mathbf{n}}$  on odd-parity conjugate variables  $\mathbf{b}^*$  (CIT, Casimir-Onsager).

#### Further insight, a particular example I

State variables  $(\mathbf{n}, \mathbf{a}, \mathbf{b})$  and entropy of the form  $S(\mathbf{n}, \mathbf{a}, \mathbf{b}) = -\sum n_j \ln n_j + Q_j n_j + \widetilde{S}(\mathbf{a}, \mathbf{b})$ . Assuming separation of dissipation potential

$$\Xi(\mathbf{n},\mathbf{a},\mathbf{b},\mathbf{n}^*,\mathbf{a}^*,\mathbf{b}^*) = \Xi^1(\mathbf{n},\mathbf{a},\mathbf{b},\mathbf{n}^*) + \Xi^2(\mathbf{n},\mathbf{a},\mathbf{b},\mathbf{a}^*,\mathbf{b}^*).$$

Then the evolution is:

$$\left( \begin{array}{c} \dot{\mathbf{n}} \\ \dot{\mathbf{a}} \\ \dot{\mathbf{b}} \end{array} \right) = \left( \begin{array}{c} \gamma \mathbf{Z}^* + \mu \mathbf{b}^* \\ \kappa \mathbf{b}^* + \nu \mathbf{Z}^* \\ -\kappa^T \mathbf{a}^* - \mu^T \mathbf{n}^* - \Xi_{\mathbf{b}^*}^2 \end{array} \right),$$

where  $\mathbf{z}^* = \mathbf{Z}^*(\mathbf{n}, \mathbf{a}, \mathbf{b}, \mathbf{n}^*, \mathbf{a}^*)$  is the solution of zero flux equation  $\gamma^T \mathbf{n}^* + \nu^T \mathbf{a}^* + \Xi_{\mathbf{z}^*} = 0$  (quasi-steady assumption).

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#### Further insight, a particular example II

Now, as  $\Xi_{z^*} = \Xi_{z^*}^1(z^*)$  we identify  $\Xi^1(z^*)$  with the dissipation potential for uncoupled chemical kinetics  $\Theta^{chem}(z^*)$ . Then

$$\Xi_{z_{\rho}^{*}}^{1} = \Theta_{z_{\rho}^{*}}^{chem} = -2 \ln \left( -\frac{z_{\rho}^{*}}{W_{\rho}} + \sqrt{\left(\frac{z_{\rho}^{*}}{W_{\rho}}\right)^{2} + 1} \right) = -\underbrace{\left(\gamma^{T} \mathbf{n}^{*} + \nu^{T} \mathbf{a}^{*}\right)_{\rho}}_{X_{\rho}}$$
$$\therefore \mathbf{Z}_{\rho}^{*}(\mathbf{n}^{*}, \mathbf{a}^{*}) = -\frac{1}{2} W_{\rho} \left( e^{1/2X_{\rho}} - e^{1/2X_{\rho}} \right)$$
$$\therefore -(\gamma \mathbf{Z}^{*}(\mathbf{n}^{*}, \mathbf{a}^{*}))_{k} = -\sum_{\rho} \gamma_{\rho}^{k} \mathbf{Z}_{\rho}^{*}(\mathbf{n}^{*}, \mathbf{a}^{*})$$

which is exactly the chemical dissipation potential  $\Theta_{n^*}^{chem}$  yielding the law of mass action within GENERIC but where "extended affinities"

$$X_{
ho} = \left( \boldsymbol{\gamma}^{\mathsf{T}} \mathbf{n}^* + \boldsymbol{
u}^{\mathsf{T}} \mathbf{a}^* 
ight)_{
ho}$$

have been introduced.

Note that  $X_{\rho}$  is independent of **b**<sup>\*</sup>. LoMA is recovered without coupling,  $\nu = 0$ ,  $\mu = 0$ .

### Further insight, a particular example III

The evolution equations may now be rewritten into the following form

$$\begin{pmatrix} \dot{\mathbf{n}} \\ \dot{\mathbf{a}} \\ \dot{\mathbf{b}} \end{pmatrix} = \begin{pmatrix} 0 & 0 & \boldsymbol{\mu} \\ 0 & 0 & \boldsymbol{\kappa} \\ -\boldsymbol{\mu}^{\mathsf{T}} & -\boldsymbol{\kappa}^{\mathsf{T}} & 0 \end{pmatrix} \begin{pmatrix} \mathbf{n}^* \\ \mathbf{a}^* \\ \mathbf{b}^* \end{pmatrix} - \begin{pmatrix} \Xi_{\mathbf{n}^*}^1 \\ \Xi_{\mathbf{a}^*}^1 \\ \Xi_{\mathbf{b}^*}^2 \end{pmatrix},$$

where

$$\Xi^{1}(\mathbf{n}, \mathbf{a}, \mathbf{b}, \mathbf{n}^{*}, \mathbf{a}^{*}) = \sum_{\rho=1}^{s} W_{\rho}(\mathbf{n}) \left( e^{-\frac{1}{2}X_{\rho}} + e^{\frac{1}{2}X_{\rho}} - 2 \right)$$

with  $X_{\rho} = \sum_{j} \gamma_{\rho}^{j} n_{j}^{*} + \sum_{i} \nu_{\rho}^{i} a_{i}^{*}$  and  $\Xi^{2}(\mathbf{n}, \mathbf{a}, \mathbf{b}, \mathbf{a}^{*}, \mathbf{b}^{*})$  being some relevant dissipation functional for description of the considered mechanical process (standard potential would be quadratic).

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# Further insight, a particular example IV Modified LoMA

The very same calculations that lead to obtaining the law of mass action within GENERIC may be applied here yielding

$$\dot{n}_{i} = \sum_{\rho} \underbrace{\left\{ \overrightarrow{k}_{\rho} \exp\left(-\frac{1}{2} \sum_{j} \nu_{\rho}^{j} a_{\rho}^{*}\right) \right\}}_{\overrightarrow{k_{\rho}}} \gamma_{\rho}^{i} \prod_{l} n_{l}^{\alpha_{\rho}^{i}} - \frac{\overrightarrow{k_{\rho}}}{\sum_{\rho}} \sum_{j} \left\{ \overleftarrow{k}_{\rho} \exp\left(\frac{1}{2} \sum_{j} \nu_{\rho}^{j} a_{j}^{*}\right) \right\}} \gamma_{\rho}^{i} \prod_{l} n_{l}^{\beta_{\rho}^{i}} + \sum_{j} \mu_{j}^{i} b_{j}^{*}, \quad i = 1, \dots, K$$

and for mechanics

$$\dot{a}_{i} = \sum_{\rho} \overrightarrow{\widetilde{k}}_{\rho} \nu_{\rho}^{i} \prod_{l} n_{l}^{\alpha_{\rho}^{l}} - \sum_{j} \overleftarrow{\widetilde{k}}_{\rho} \nu_{\rho}^{i} \prod_{l} n_{l}^{\beta_{\rho}^{l}} + \sum_{j} \kappa_{j}^{i} b_{j}^{*}, \quad i = 1, \dots, m_{1}$$
$$\dot{b}_{i} = -\sum_{j=1}^{K} \mu_{j}^{i} n_{j}^{*} - \sum_{k=1}^{m_{1}} \kappa_{k}^{i} a_{k}^{*} - \Xi_{b_{i}^{*}}^{2} (\mathbf{n}, \mathbf{a}, \mathbf{b}, \mathbf{a}^{*}, \mathbf{b}^{*}), \quad i = 1, \dots, m_{2}$$

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IWNET 12 20 / 25

# Customer of thermodynamics: Model of BR, algorithm

- Identification of crucial biochemical processes
- Modified LoMA
- Parameter setting
- FEM implementation



#### Femur in a healthy individual, simulation



Standard serum levels of considered biochemical factors corresponding to a *healthy state/individual*: estradiol 50  $\frac{pg}{ml}$ , PTH 34  $\frac{pg}{ml}$ , RANKL 46.2  $\frac{pg}{ml}$ , OPG 36  $\frac{pg}{ml}$ , and NO levels correspond to intake of  $0.044 \frac{mg}{kg}$  of nitroglycerin per day. Number of loading cycles is N = 10,000 also corresponding to healthy mechanical stimulus per day

#### Predictions and observations

Comparison of prediction of BMD changes from mechano-chemical (CIT) model to data found in literature.

	hyperPTH	post-menopausal osteoporosis	estradiol 17.5 <u>pg</u>	running 6 mi/day
BMD <sub>ind</sub>	-10.4%	-5.6%	<-1%	+16.2%
literature	-(7-15)%	-5.7%	<-1%	+14.1%

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#### Conclusions

- Can coupling analysis be carried out outside of linear regime (close to equilibrium) when majority of chemical processes are considered to be far from equilibrium? How?
- (partners) Multi-disciplinary appoach is needed for many problems non-equilibrium thermodynamics can be used with success
- (partners&customers) Bridging of this gap with non-equilibrium thermodynamics
- Experimental verification of mechano-chemical coupling

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## A nonlinear coupling of chemical kinetics with mechanics

#### V. Klika<sup>1,2</sup>, M. Grmela<sup>3</sup>

<sup>1</sup>Institute of Thermomechanics of the AS CR, CR klika@it.cas.cz
<sup>2</sup>Dept. of Mathematics, FNSPE, Czech Technical University in Prague, CR
<sup>3</sup>Chemical Engineering, Ecole Polytechnique, Montreal, Canada

#### 23.8.12 Roros, IWNET 12