

# THE EFFECT OF NANOPARTICLE HYDROPHOBICITY ON THE RHEOLOGY OF HIGHLY CONCENTRATED EMULSIONS

N.N. TSHILUMBU\*, I. MASALOVA

Department of Civil Engineering, Faculty of Engineering, Cape Peninsula University of Technology,  
PO Box 652, Cape Town 8000, Republic of South Africa

\* Corresponding author: [tshilumbun@cput.ac.za](mailto:tshilumbun@cput.ac.za)  
Fax: x27.21.4603990

Received: 18.2.2013, Final version: 7.8.2013

## ABSTRACT:

A series of fumed silica nanoparticles were used as an additional emulsifier for highly concentrated (HC) water-in-oil (W/O) emulsions. These nanoparticles, with different hydrophobicity index (HI) in the  $0.60 - 1.34$  and  $HI > 3$  range, were mixed with the conventional low molecular weight surfactant, sorbitan monooleate (SMO), in the oil phase prior to the emulsification process. The rheological properties of these emulsions were measured and compared with the properties of emulsions stabilized with SMO alone. In the mixed emulsifier system, the changes in rheological parameters were clearly expressed as a function of HI. The mixture of silica nanoparticles and SMO significantly increases the yield stress and plateau modulus of fresh emulsion, compared to the SMO only system. The effect was found to be more pronounced with a decrease in the HI. This is probably related to the reduction in micelle content with the decrease in HI, owing to a concomitant increase in the amount of SMO adsorbed onto the particle surface. Then, interestingly, the Foudazi-Masalova model recently developed for surfactant-stabilized highly concentrated emulsions (HCE) was found to describe successfully the rheological behavior of emulsions in the presence of a mixture of surfactant and fumed nanosilica.

## ZUSAMMENFASSUNG:

Mehrere pyrogene Silika-Nanopartikel wurden als zusätzliche Emulgierungsmittel für hochkonzentrierte (HC) Wasser-in-Öl (W/O)-Emulsionen verwendet. Diese Nanopartikel, die unterschiedliche Hydrophobizitätsindizes (HI) im Bereich von 0.60 bis 1.34 und  $HI > 3$  besitzen, wurden mit einem konventionellen, niedermolekularen Surfactanten (Sorbitanmonooleat, SMO) in die Ölphase vor dem Emulsifikationsprozess gemischt. Die rheologischen Eigenschaften dieser Emulsionen wurden gemessen und mit den Eigenschaften der mit dem SMO allein stabilisierten Emulsionen verglichen. Bei den gemischten Emulgierungsmitteln waren die rheologischen Eigenschaften eindeutig vom HI-Wert abhängig. Die Mischungen der Silika-Nanopartikel und dem SMO erhöhten in signifikanter Weise die Fließspannung und den Plateau-Modul der frischen Emulsion im Vergleich zu dem System, das mit SMO allein stabilisiert wurde. Der Effekt trat deutlicher bei einem geringeren HI-Wert auf. Dies steht wahrscheinlich mit der Abnahme des Mizellengehaltes aufgrund der als Begleiterscheinung auftretenden Zunahme des SMO-Gehalts in Zusammenhang, das auf der Partikeloberfläche adsorbiert ist. Weiterhin beschreibt interessanterweise das Foudazi-Masalova-Modell, das kürzlich für mit durch Surfaktanten-stabilisierte hochkonzentrierte Emulsionen (HCE) entwickelt wurde, das rheologische Verhalten der Emulsionen mit Surfaktanten und pyrogenen Nanosilikapartikeln.

## RÉSUMÉ:

Une série de nanoparticules de fumée de silice a été utilisée comme émulsifiant additionnel d'émulsions eau-dans-huile (W/O) concentrées. Ces nanoparticules, possédant différents index d'hydrophobicité (HI) dans la gamme  $0.6 - 1.34$  et  $HI > 3$ , ont été mélangées avec du surfactant conventionnel de bas poids moléculaire, du mono oléate de sorbitane (SMO), dans une phase d'huile, avant le procédé émulsifiant. Les propriétés rhéologiques de ces émulsions ont été mesurées et comparées avec celles des émulsions stabilisées seulement par le SMO. Dans le système d'émulsifiant mixte, les changements des paramètres rhéologiques avec le HI sont clairement révélés. La mixture de nanoparticules et de SMO augmente significativement la contrainte seuil et le module plateau de l'émulsion fraîche, comparé au système avec le SMO seul. L'effet s'est avéré plus prononcé lorsque le HI décroît. Ceci est probablement lié à la réduction du nombre de micelles avec la baisse du HI, du à l'augmentation correspondante de la quantité de SMO adsorbé sur la surface de la particule. Ensuite, de manière intéressante, le modèle Foudazi-Masalova récemment développé pour les émulsions concentrées stabilisées par du surfactant (HCE), décrit avec succès le comportement rhéologique en présence du mélange de surfactant et de fumée de silice.

**KEY WORDS:** emulsions, rheology, fumed nano-silica, surfactant

© Appl. Rheol. 23 (2013) 62835

DOI: 10.3933/ApplRheol-23-62835

This is an extract of the complete reprint-pdf, available at the Applied Rheology website

<http://www.appliedrheology.org>

- [13] Binks BP, Desforges A, Duff DG: Synergistic stabilization of emulsions by a mixture of surface-active nanoparticles and surfactants, *Langmuir* 23 (2007) 1098–1106.
- [14] Tshilumbu NN, Kharatyan E, Masalova I: Effect of nanoparticle hydrophobicity on stability of highly concentrated emulsions, *J. Dispers. Sci. Technol.* Accepted for publication.
- [15] Simovic S, Prestidge CA: Nanoparticle layers controlling drug release from emulsions, *Eur. J. Pharm. Biopharm.* 67 (2007) 39–47.
- [16] Windhab E: Bericht der IV. Tagung der Lebensmittelrheologie, Detmold (1993).
- [17] Foudazi R, Masalova I, Malkin AY: Effect of interdroplet interaction on elasticity of highly concentrated emulsions, *Appl. Rheol.* 20 (2010) 45096–45106.
- [18] Foudazi R, Masalova I, Malkin AY: Flowbehavior of highly concentrated emulsions of supersaturated aqueous solution in oil, *Rheol. Acta* 50 (2011) 897–907.
- [19] Reynolds AP, Gilbert PE, White WJ: High internal phase water-in-oil emulsions studied by small-angle-scattering, *Phys. Chem. B* 104 (2000) 7012–7022.
- [20] Reynolds PA, Gilbert PE, White WJ: High internal phase water-in-oil emulsions and related microemulsions studied by small angle neutron scattering, *J. Phys. Chem. B* 105 (2001) 6925–6932.
- [21] Zank J, Reynolds PA, Jackson AJ, Baranyai KJ, Perriman AW, Barker JG, Kim M-H, White WJ: Aggregation in a high internal phase emulsion observed by SANS and USANS, *Physica B* 385–386 (2006) 776–779.
- [22] Espinosa CE, Guo Q, Singh V, Behrens SH: Particle charging and charge screening in nonpolar dispersions with nonionic surfactants, *Langmuir* 26 (2010) 16941–16948.
- [23] Hsu MF, Dufresne ER, Weitz DA: Charge stabilization in nonpolar solvents, *Langmuir* 21 (2005) 4881–4887.
- [24] Reynolds PA, Henderson MJ, Zank J, White JW: Complex layering observed in high internal phase emulsions at a silicon surface by neutron reflectometry, *J. Colloid Interface Sci.* 364 (2011) 539–545.
- [25] Arditty S, Schmitt V, Lequeux F, Leal-Calderon F: Interfacial properties in solid-stabilized emulsions, *Eur. Phys. J. B* 4 (2005) 381–393.
- [26] Dukhin AS, Goetz PJ: How non-ionic “electrically neutral” surfactants enhance electrical conductivity and ion stability in non-polar liquids, *J. Electroanal. Chem.* 588 (2006) 44–55.
- [27] Poovarodom S, John CB: Effect of particle and surfactant acid-base properties on charging of colloids in a polar media, *J. Colloid Interface Sci.* 346 (2010) 370–377.
- [28] Menner A, Ikem V, Salgueiro M, Bismarck A: High internal phase emulsion templates solely stabilised by functionalised titania nanoparticles, *Chem. Comm.* (2007) 4274–4276.
- [29] Richard HH, Don HC, Heldon KH, Scott BP: Emulsion phase having improved stability, US Patent (2004) 6808573.
- [30] Princen HM, Kiss AD: Rheology of foams and highly concentrated emulsions. A: An experimental study of the shear viscosity and yield stress of concentrated emulsions, *J. Colloid Interface Sci.* 128 (1989) 176–187.
- [31] Simovic S, Prestidge CA: Nanoparticles of varying hydrophobicity at the emulsion droplet-water interface: adsorption and coalescence stability, *Langmuir* 20 (2004) 8357–8365.
- [32] Smitha S, Shajesh P, Mukundan P, Warrier KGK: Synthesis of mesoporous hydrophobic silica microspheres through a modified sol-emulsion-gel process, *J. Sol-Gel Sci. Technol.* 48 (2008) 356–361.
- [33] Masalova I, Foudazi R, Malkin, AY: The rheology of highly concentrated emulsions stabilized with different surfactants, *Colloids and Surfaces A: Physicochem. Eng. Aspects* 375 (2011) 76–86.
- [34] Yates DE, Dack SW: Determination of emulsion explosives with Span-80 as emulsifier, US Patent (1987) 4710248.
- [35] Sudweeks W: Physical and Chemical Properties of Industrial Slurry Explosives, *Ind. Eng. Chem. Prod. Res. Dev.* 24 (1985) 432–436.
- [36] White WJ, Henderson JM, Perriman A: The crystallization of supersaturated emulsions, *Isis Experimental Report* (2004): 14591.
- [37] Binks BP, Desforges A, Duff DG: Synergistic Stabilization of Emulsions by a Mixture of Surface-Active Nanoparticles and Surfactant, *Langmuir* 23 (2007) 1098–1106.
- [38] Midmore BR: Preparation of a novel silica-stabilized oil/water emulsion, *Colloids and Surfaces A: Physicochem. Eng. Aspects* 132 (1998) 257–265.
- [39] Midmore BR: Effect of aqueous phase composition on the properties of a silica-stabilized W/O emulsion, *J. Coll. Interf. Sci.* 213 (1999) 352–359.
- [40] Tambe DE, Sharma MM: Factors controlling the stability of colloid-stabilized emulsions. I. An experimental investigation, *J. Coll. Interf. Sci.* 157 (1993) 244–53.
- [41] Tambe DE, Sharma MM: Factors controlling the stability of colloid-stabilized emulsions. II. A model for the rheological properties of colloid-Laden interface, *J. Coll. Interf. Sci.* 162 (1994) 1–10.
- [42] Tambe DE, Sharma MM: The effect of colloidal particles on fluid-fluid interfacial properties and emulsion stability, *Adv Coll. Interf. Sci.* 52 (1994) 1–65.
- [43] Midmore, BR: Synergy between silica and polyoxyethylene surfactants in the Formation of O/W Emulsions, *Colloids and Surf. A: Physicochem. Eng. Aspects* 145 (1998) 133–143.
- [44] Binks BP, Philip J, Rodrigues JA: Inversion of Silica-Stabilized Emulsions Induced by Particle Con-

This is an extract of the complete reprint-pdf, available at the Applied Rheology website  
<http://www.appliedrheology.org>

This is an extract of the complete reprint-pdf, available at the Applied Rheology website  
<http://www.appliedrheology.org>

Applied Rheology  
 Volume 23 · Issue 6

62835-10

- centration, *Langmuir* 21 (2005) 3296–3302.
- [45] Khan SA, Zoeller NJ: Dynamic Rheological Behavior of Flocculated Fumed Silica Suspensions, *J. Rheol.* 37 (1993) 1225–1235.
- [46] Raghavan SR, Walls HJ, Khan SA: Colloidal Interactions between Particles with Tethered Nonpolar Chains Dispersed in Polar Media: Direct Correlation between Dynamic Rheology and Interaction Parameters, *Langmuir* 16 (2000) 7920.
- [47] Barthel H: Surface interactions of dimethylsiloxy group-modified fumed silica, *Colloids Surf. A* 101 (1995) 217.
- [48] Binks BP, Clint JH, Whitby CP: Rheological Behavior of Water-in-Oil Emulsions Stabilized by Hydrophobic Bentonite Particles, *Langmuir* 21 (2005) 5307–5316.
- [49] Lagaly G, Reese M, Abend S: Smectites as colloidal stabilizers of emulsions – II. Rheological properties of smectite-laden emulsions, *Appl. Clay Sci.* 14 (1999) 279.
- [50] Gosa KL, Uricanu V: Emulsions stabilized with PEOPPO-PEO block copolymers and silica, *Colloids and Surf. A: Physicochem. Eng. Aspects* 197 (2002) 257–269.



This is an extract of the complete reprint-pdf, available at the Applied Rheology website  
<http://www.appliedrheology.org>